

## Synthesis of Allyl Aryl Sulfone Derivatives from *Baylis–Hillman* Acetates in Water

by Konkala Karnakar, Jilla Shankar, Sabbavarapu Narayana Murthy, and  
Yadavalli Venkata Durga Nageswar\*

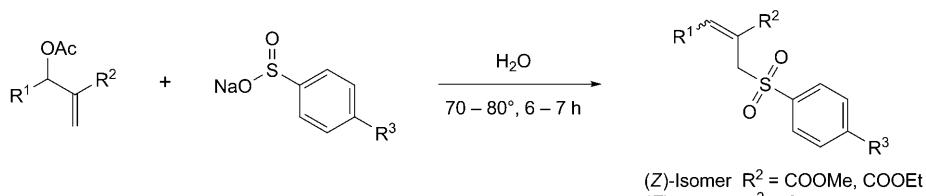
Organic Chemistry Division-I, Indian Institute of Chemical Technology, Hyderabad 500607, India  
(phone: +91-40-27191654; e-mail: dryvdnageswar@gmail.com)

Various phenyl and *p*-tolyl allyl sulfone derivatives were prepared stereoselectively by reacting *Baylis–Hillman* acetates with sodium 4-R-benzenesulfinate ( $R = H, Me$ ) in  $H_2O$ . The reaction was very efficient in providing the corresponding sulfone derivatives in good to excellent yields (*Table*).

**Introduction.** – As a part of our ongoing programme initiated towards the development of ecological chemical methodologies [1], we became interested in *Baylis–Hillman* acetates as reactive intermediates. Herein, we report the synthesis of disubstituted phenyl or *p*-tolyl allyl sulfone derivatives in  $H_2O$  under catalyst-free conditions.

**Results.** – In the course of the present work, *Baylis–Hillman* acetates were treated with sodium benzenesulfonates in  $H_2O$  at room temperature. The reaction resulted in the corresponding allyl sulfone derivatives in 71% yield. However, the same reaction in  $H_2O$  under heating proceeded in a much better way. After having optimized the reaction conditions, *Baylis–Hillman* acetates and sodium 4-R-benzenesulfonates ( $R = H, Me$ ) in  $H_2O$  were heated at 70–80° to give the products in excellent yields of up to 94% within 6 h (*Scheme; Table, Entry 1*). Encouraged by the result, we first prepared various *Baylis–Hillman* acetates by reported literature procedures from commercially available aldehydes  $R^1\text{--CHO}$  and *Michael* acceptors such as methyl or ethyl acrylate ( $\text{CH}_2=\text{CHCOOR}$ ) or acrylonitrile ( $\text{CH}_2=\text{CHCN}$ ) in the presence of DABCO (=1,4-

Scheme. *Synthesis of Allyl Aryl Sulfone Derivatives from Baylis–Hillman Acetates and Sodium Benzenesulfonates*



$R^1 = \text{aromatic, heteroaromatic, aliphatic}$   
 $R^2 = \text{COOMe, COOEt, CN}$   
 $R^3 = H, Me$

Table. *Synthesis of Allyl Phenyl or Allyl p-Tolyl Sulfone Derivatives from Baylis–Hillman Acetates and Sodium Benzenesulfinate or Sodium 4-Methylbenzenesulfinate<sup>a</sup>*

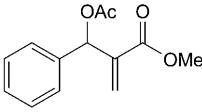
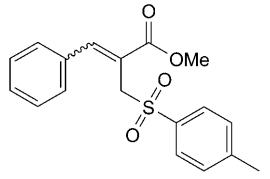
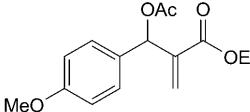
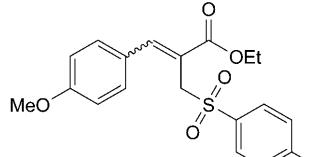
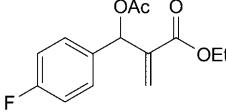
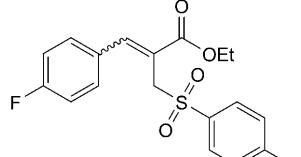
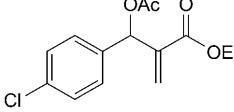
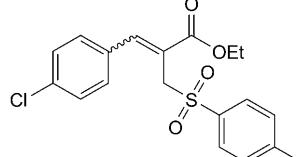
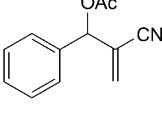
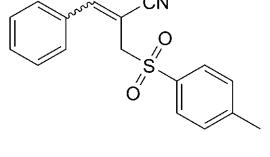
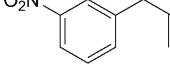
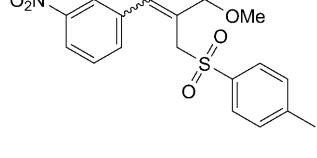
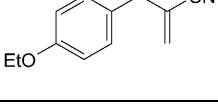
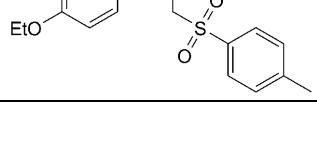
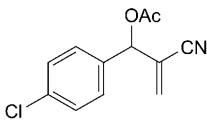
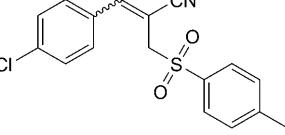
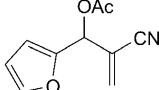
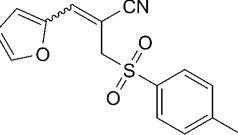
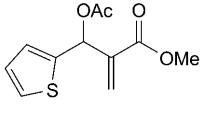
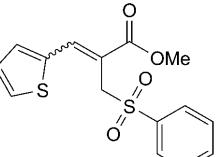
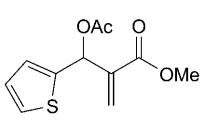
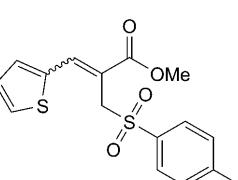
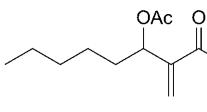
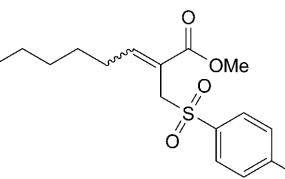
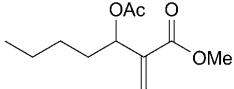
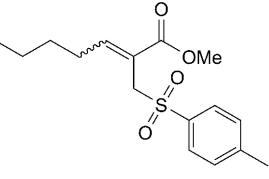
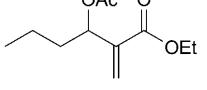
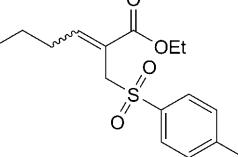
Entry	Substrate	Product	Time [h]	Yield [%] <sup>b</sup>	(E)/(Z) <sup>c</sup>
1			6	94	2:98
2			7	92	3:97
3			6	91	3:97
4			6	91	3:97
5			6	87	95:5
6			6	95	4:96
7			6	91	95:5

Table (cont.)

Entry	Substrate	Product	Time [h]	Yield [%] <sup>b</sup>	(E)/(Z) <sup>c</sup>
8			6	89	95:5
9			7	84	95:5
10			6	89	3:97
11			6	87	3:97
12			7	74	5:95
13			7	73	5:95
14			7	73	4:96

<sup>a</sup>) Reaction conditions: Baylis–Hillman acetate (1.0 mmol), sodium sulfinate (1.2 mmol), H<sub>2</sub>O (15 ml), 70–80°, 6–7 h. <sup>b</sup>) Yield of isolated sulfone derivatives. <sup>c</sup>) Selectivity was determined by <sup>1</sup>H-NMR analysis.

diazabicyclo[2.2.2]octane) as catalyst and Ac<sub>2</sub>O as acetylating agent. The resulting adducts were then treated with sodium benzenesulfinate or sodium 4-methylbenzenesulfinate (*Table*). In the present study, aldehydes R<sup>1</sup>-CHO with electron-donating groups gave the corresponding allyl sulfone derivatives in lower yields in comparison to aldehydes with electron-withdrawing groups. In the case of acetates of aliphatic *Baylis–Hillman* adducts, the corresponding allyl sulfone derivatives were also obtained in lower yields (*Table, Entries 12–14*). The (2Z)-configuration of the 2-(sulfonylmethyl)acrylates were assigned on the basis of <sup>1</sup>H- and <sup>13</sup>C-NMR shifts in comparison with literature values [2]. In the case of *Baylis–Hillman* adducts prepared from acrylonitrile, the reaction with sodium 4-methylbenzenesulfinate gave the corresponding 2-(sulfonylmethyl)acrylonitriles with reversal of selectivity, *i.e.*, the (2E)-isomers were the major product. This fact was also confirmed by <sup>1</sup>H- and <sup>13</sup>C-NMR measurements and by comparison with reported values.

In conclusion, we developed an elegant protocol for the synthesis of allyl sulfone derivatives in high yields by reaction of *Baylis–Hillman* acetates with sodium 4-R-benzenesulfinates (R = H, Me) in H<sub>2</sub>O. This practical method may have widespread applications in the context of ecologically oriented organic synthesis.

*K. K., J. S., and S. N. M.* are grateful to the *Council of Scientific and Industrial Research (CSIR)*, New Delhi, India, for providing fellowships.

### Experimental Part

*General.* TLC: Precoated silica-gel plates (60 F<sub>254</sub>, 0.2-mm layer, *E. Merck*). Column chromatography (CC): Silica gel, 60–120 mesh. M.p.: *Fischer–Johns* apparatus; uncorrected. IR Spectra: *Thermo Nicolet Nexus 670* FT-IR spectrophotometer; in KBr;  $\tilde{\nu}$  in cm<sup>-1</sup>. <sup>1</sup>H- and <sup>13</sup>C-NMR Spectra: *Bruker Avance 300*, *Innova 400* MHz and *Bruker Gemini* 200 MHz instrument; in CDCl<sub>3</sub>;  $\delta$  in ppm rel. to Me<sub>4</sub>Si as internal standard, *J* in Hz. ESI-MS: *Finnigan MAT 1020* mass spectrometer; in *m/z*.

*Allyl Aryl Sulfone Derivatives: General Procedure.* To a flask containing the *Baylis–Hillman* acetate (1 mmol) in H<sub>2</sub>O (15 ml), the sodium benzenesulfinate (1.2 mmol) was added. The suspension was magnetically stirred at 70–80° until the reaction was complete (TLC monitoring). The mixture was then extracted with AcOEt (4 × 10 ml), the extract washed with H<sub>2</sub>O and brine dried (anh. Na<sub>2</sub>SO<sub>4</sub>), and concentrated, and the resulting crude product purified by CC (AcOEt/hexane 1:9): substituted allyl sulfone derivative in yields up to 94% (*Table*).

*Methyl (2Z)-2-[(4-Methylphenyl)sulfonyl]methyl]-3-phenylprop-2-enoate* (*Table, Entry 1*): Light brown solid. M.p. 120°. IR: 2982, 1713, 1317, 1266, 1141. <sup>1</sup>H-NMR: 7.91 (s, 1 arom. H); 7.74 (d, *J* = 8.3, 2 arom. H); 7.52 (d, *J* = 7.5, 2 arom. H); 7.38 (m, 3 arom. H); 7.29 (d, *J* = 7.5, 2 arom. H); 4.41 (s, CH<sub>2</sub>SO<sub>2</sub>); 3.61 (s, MeO); 2.46 (s, Me). <sup>13</sup>C-NMR: 166.8; 146.0; 144.6; 136.1; 133.5; 129.5; 129.0; 128.6; 128.4; 120.9; 55.0; 52.2; 21.5. ESI-MS: 353 ([M + Na]<sup>+</sup>).

*Ethyl (2Z)-3-(4-Methoxyphenyl)-2-[(4-methylphenyl)sulfonyl]methyl]prop-2-enoate* (*Table, Entry 2*): Yellow oil. IR: 2987, 1713, 1610, 1510, 1319, 1144. <sup>1</sup>H-NMR: 7.89 (s, 1 H); 7.76 (d, *J* = 8.1, 2 arom. H); 7.58 (d, *J* = 8.6, 2 arom. H); 7.28 (d, *J* = 8.8, 2 arom. H); 6.92 (d, *J* = 8.8, 2 arom. H); 4.50 (s, CH<sub>2</sub>SO<sub>2</sub>); 4.00 (q, *J* = 7.1, MeCH<sub>2</sub>O); 3.84 (s, MeO); 2.42 (s, Me); 1.20 (t, *J* = 7.1, MeCH<sub>2</sub>O). <sup>13</sup>C-NMR: 166.7; 160.9; 145.2; 131.5; 129.5; 128.6; 126.2; 114.1; 61.3; 55.4; 55.3; 21.5; 14.0. ESI-MS: 397 ([M + Na]<sup>+</sup>).

*Ethyl (2Z)-3-(4-Fluorophenyl)-2-[(4-methylphenyl)sulfonyl]methyl]prop-2-enoate* (*Table, Entry 3*): White solid. M.p. 65°. IR: 2984, 1710, 1600, 1508, 1317, 1142. <sup>1</sup>H-NMR: 7.86 (s, 1 H); 7.71 (d, *J* = 7.1, 2 H); 7.54 (t, *J* = 6.7, 2 H); 7.27 (d, *J* = 7.7, 2 H); 7.05 (t, *J* = 8.4, 2 H); 4.41 (s, 2 H); 4.02 (q, *J* = 6.9, 2 H); 2.41 (s, 3 H); 1.22 (t, *J* = 6.9, 3 H). <sup>13</sup>C-NMR: 164.8; 161.5; 144.5; 132.5; 131.3; 131.2; 129.7; 129.4; 128.3; 121.0; 115.7; 115.4; 61.2; 54.8; 21.4; 14.0. ESI-MS: 385 ([M + Na]<sup>+</sup>).

*Ethyl (2Z)-3-(4-Chlorophenyl)-2-[(4-methylphenyl)sulfonyl]methyl]prop-2-enoate (Table, Entry 4):* Colorless oil. IR: 2985, 1715, 1315, 1268, 1131.  $^1\text{H-NMR}$ : 7.85 (s, 1 H); 7.70 (d,  $J = 8.3$ , 2 H); 7.44 (d,  $J = 9.0$ , 2 H); 7.24–7.33 (m, 4 H); 4.44 (s, 2 H), 4.06 (q,  $J = 6.7$ , 2 H); 2.40 (s, 3 H); 1.22 (t,  $J = 6.7$ , 3 H).  $^{13}\text{C-NMR}$ : 166.1; 144.9; 136.2; 132.1; 130.5; 129.6; 128.9; 128.4; 121.7; 61.6; 54.9; 21.5; 14.0. ESI-MS: 379 ( $[M + \text{H}]^+$ ).

*(2E)-2-[(4-Methylphenyl)sulfonyl]methyl]-3-phenyl-prop-2-enenitrile (Table, Entry 5):* Light brown solid. M.p. 90°. IR: 2212, 1623, 1311, 1146.  $^1\text{H-NMR}$ : 7.81 (d,  $J = 8.3$ , 2 H); 7.55 (s, 1 H); 7.34–7.40 (m, 7 H); 4.17 (s, 2 H); 2.46 (s, 3 H).  $^{13}\text{C-NMR}$ : 152.0; 132.2; 130.5; 130.1; 129.0; 128.9; 128.7; 118.1; 103.2; 56.4; 21.7. ESI-MS: 320 ( $[M + \text{Na}]^+$ ).

*Methyl (2Z)-2-[(4-Methylphenyl)sulfonyl]methyl]-3-(3-nitrophenyl)prop-2-enoate (Table, Entry 6):* Light yellow solid. M.p. 110°. IR: 2982, 1713, 1317, 1266, 1141.  $^1\text{H-NMR}$ : 8.20 (d,  $J = 8.3$ , 2 H); 7.94 (d,  $J = 7.8$ , 1 H); 7.91 (s, 1 H); 7.71 (d,  $J = 7.8$ , 2 H); 7.60 (t,  $J = 7.8$ , 1 H); 7.28 (d,  $J = 7.8$ , 2 H); 4.33 (s, 2 H); 3.69 (s, 3 H); 2.43 (s, 3 H).  $^{13}\text{C-NMR}$ : 166.5; 145.3; 140.3; 138.4; 135.6; 133.7; 129.5; 128.3; 127.4; 123.2; 56.5; 52.2; 21.2. ESI-MS: 376 ( $[M + \text{H}]^+$ ).

*(2E)-3-(4-Ethoxyphenyl)-2-[(4-methylphenyl)sulfonyl]methyl]prop-2-enenitrile (Table, Entry 7):* Yellow oil. IR: 2927, 2210, 1627, 1321, 1148.  $^1\text{H-NMR}$ : 7.54 (d,  $J = 8.6$ , 2 H); 7.40 (d,  $J = 6.7$ , 2 H); 7.26 (t,  $J = 6.7$ , 2 H); 6.81 (d,  $J = 8.6$ , 2 H); 6.54 (s, 1 H), 4.01 (q,  $J = 6.7$ , 2 H); 3.69 (s, 2 H); 2.43 (s, 3 H); 1.41 (t,  $J = 6.7$ , 3 H).  $^{13}\text{C-NMR}$ : 160.4; 144.0; 138.4; 133.2; 130.5; 129.3; 128.5; 127.4; 114.2; 104.2; 63.7; 57.2; 21.3; 14.6. ESI-MS: 364 ( $[M + \text{Na}]^+$ ).

*(2E)-3-(4-Chlorophenyl)-2-[(4-methylphenyl)sulfonyl]methyl]prop-2-enenitrile (Table, Entry 8):* White solid. M.p. 110°. IR: 2928, 2212, 1627, 1313, 1144.  $^1\text{H-NMR}$ : 7.80 (d,  $J = 8.1$ , 1 H); 7.38 (m, 6 H); 7.30 (d,  $J = 8.3$ , 2 H); 4.42 (s, 2 H); 2.47 (s, 3 H).  $^{13}\text{C-NMR}$ : 150.8; 145.2; 136.2; 134.9; 131.4; 130.7; 130.1; 129.3; 129.1; 128.6; 118.0; 103.4; 56.3; 21.7. ESI-MS: 354 ( $[M + \text{Na}]^+$ ).

*(2E)-3-(Furan-2-yl)-2-[(4-methylphenyl)sulfonyl]methyl]prop-2-enenitrile (Table, Entry 9):* Brown solid. M.p. 125°. IR: 2925, 2212, 1623, 1311, 1146.  $^1\text{H-NMR}$ : 7.78 (d,  $J = 8.1$ , 2 H); 7.57 (s, 1 H); 7.38 (d,  $J = 8.1$ , 2 H); 7.04 (d,  $J = 3.5$ , 1 H); 6.96 (s, 1 H); 6.54 (s, 1 H); 3.98 (s, 2 H); 2.45 (s, 3 H).  $^{13}\text{C-NMR}$ : 148.7; 145.8; 137.1; 134.5; 130.0; 128.5; 119.6; 117.1; 112.7; 112.2; 60.5; 21.6. ESI-MS: 305 ( $[M + \text{NH}_3]^+$ ).

*Methyl (2Z)-2-[(Phenylsulfonyl)methyl]-3-(thiophen-2-yl)prop-2-enoate (Table, Entry 10):* Light brown oil. IR: 2984, 1710, 1600, 1508, 1317, 1142.  $^1\text{H-NMR}$ : 8.00 (s, 1 H); 7.78 (d,  $J = 8.3$ , 2 H); 7.57 (d,  $J = 3.7$ , 1 H); 7.52 (d,  $J = 5.2$ , 1 H); 7.30 (d,  $J = 8.3$ , 3 H); 7.12 (t,  $J = 3.7$ , 1 H); 4.55 (s, 2 H); 3.54 (s, 3 H).  $^{13}\text{C-NMR}$ : 166.4; 139.0; 137.5; 134.2; 133.5; 131.5; 129.3; 128.5; 128.2; 62.4; 52.7. ESI-MS: 345 ( $[M + \text{Na}]^+$ ).

*Methyl (2Z)-2-[(4-Methylphenyl)sulfonyl]methyl]-3-(thiophen-2-yl)prop-2-enoate (Table, Entry 11):* Light brown oil. IR: 2984, 1710, 1600, 1508, 1317, 1142.  $^1\text{H-NMR}$ : 8.00 (s, 1 H); 7.78 (d,  $J = 8.3$ , 2 H); 7.57 (d,  $J = 3.7$ , 1 H); 7.52 (d,  $J = 5.2$ , 1 H); 7.30 (d,  $J = 8.3$ , 2 H); 7.12 (t,  $J = 3.7$ , 1 H); 4.55 (s, 2 H); 3.54 (s, 3 H); 2.45 (s, 3 H).  $^{13}\text{C-NMR}$ : 166.4; 139.0; 137.5; 134.2; 133.5; 131.5; 129.3; 128.5; 128.2; 62.4; 52.7; 21.4. ESI-MS: 359 ( $[M + \text{Na}]^+$ ).

*Methyl (2Z)-2-[(4-Methylphenyl)sulfonyl]methyl]oct-2-enoate (Table, Entry 12):* Colorless oil. IR: 2953, 1720, 1321, 1143, 1085.  $^1\text{H-NMR}$ : 7.72 (d,  $J = 8.3$ , 2 H); 7.32 (d,  $J = 8.3$ , 2 H); 7.09 (t,  $J = 7.5$ , 1 H); 4.22 (s, 2 H); 3.05 (s, 3 H); 2.42 (s, 3 H); 2.14 (q,  $J = 6.7$ , 2 H); 2.0 (m, 2 H); 1.32 (m, 4 H); 0.87 (t,  $J = 7.5$ , 3 H).  $^{13}\text{C-NMR}$ : 165.7; 151.3; 144.3; 131.5; 129.2; 128.2; 120.2; 60.2; 53.6; 29.9; 28.7; 22.7; 21.9; 21.1; 13.7. ESI-MS: 347 ( $[M + \text{Na}]^+$ ).

*Methyl (2Z)-2-[(4-Methylphenyl)sulfonyl]methyl]hept-2-enoate (Table, Entry 13):* Colorless oil. IR: 2956, 1720, 1318, 1144, 1084.  $^1\text{H-NMR}$ : 7.72 (d,  $J = 8.3$ , 2 H); 7.32 (d,  $J = 8.3$ , 2 H); 7.09 (t,  $J = 7.5$ , 1 H); 4.22 (s, 2 H); 3.05 (s, 3 H); 2.42 (s, 3 H); 2.14 (q,  $J = 6.7$ , 2 H); 1.32 (m, 4 H); 0.87 (t,  $J = 7.5$ , 3 H).  $^{13}\text{C-NMR}$ : 165.7; 151.3; 144.5; 131.5; 129.2; 128.2; 120.2; 60.2; 53.6; 29.9; 28.7; 21.9; 21.1; 13.7. ESI-MS: 333 ( $[M + \text{Na}]^+$ ).

*Ethyl (2Z)-2-[(4-Methylphenyl)sulfonyl]methyl]hex-2-enoate (Table, Entry 14):* Colorless oil. IR: 2948, 1720, 1319, 1146, 1087.  $^1\text{H-NMR}$ : 7.68 (d,  $J = 6.9$ , 2 H); 7.28 (d,  $J = 6.9$ , 2 H); 7.03 (t,  $J = 7.5$ , 1 H); 4.15 (s, 2 H); 3.91 (q,  $J = 7.1$ , 2 H); 2.42 (s, 3 H); 2.20 (q,  $J = 7.5$ , 2 H); 1.46 (m, 2 H); 1.13 (t,  $J = 7.1$ , 3 H); 0.94 (t,  $J = 7.1$ , 3 H).  $^{13}\text{C-NMR}$ : 165.0; 150.4; 136.4; 129.3; 128.8; 128.7; 60.6; 53.9; 31.8; 31.3; 21.5; 14.0; 13.8. ESI-MS: 333 ( $[M + \text{Na}]^+$ ).

## REFERENCES

- [1] S. N. Murthy, B. Madhav, V. P. Reddy, K. R. Rao, Y. V. D. Nageswar, *Tetrahedron Lett.* **2009**, *50*, 5009.
- [2] D. Basavaiah, K. Muthukumaran, B. Sreenivasulu, *Synthesis* **2000**, 545; P. H. Manson, N. D. Esmile, *Tetrahedron* **1994**, *41*, 12001; P. Shanmugam, P. R. Singh, *Synlett* **2001**, 1314; D. Basavaiah, P. K. S. Sarma, N. Kumaragurubaran, *J. Org. Chem.* **1999**, *64*, 1197; T. Funabiki, H. Hosomi, S. Yoshida, K. Tarama, *J. Am. Chem. Soc.* **1982**, *104*, 1560; A. Foucaud, F. El Guemmout, *Bull. Soc. Chim. Fr.* **1989**, 403.

Received September 6, 2010